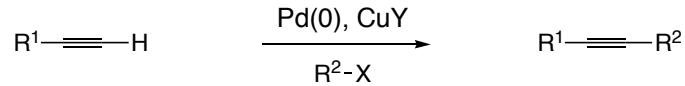


**Réaction de Sonogashira**

$\text{R}^1$  = alkyl, vinyl, allyl, aryl (alcynes terminaux)

$\text{R}^2$  = vinyl, allyl, aryl

X = Br, I, OTf

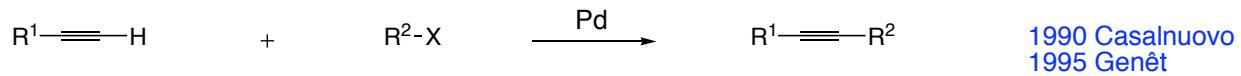
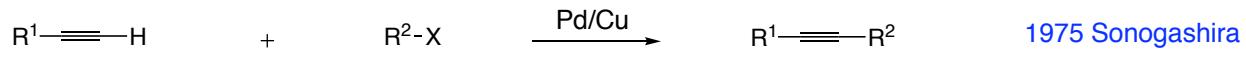
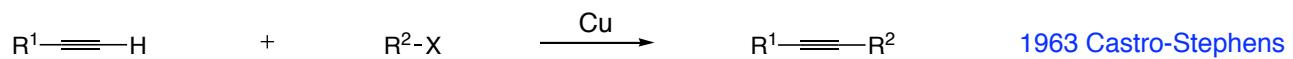
Y = I, Br

Sonogashira 75TL4467

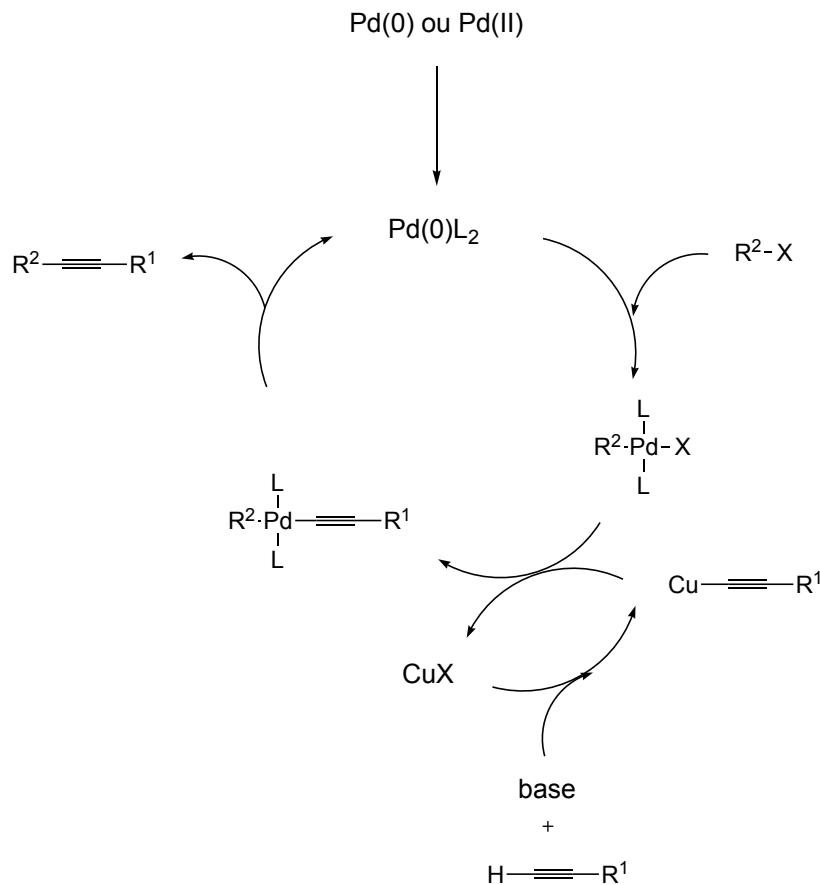
Revues

91COC521

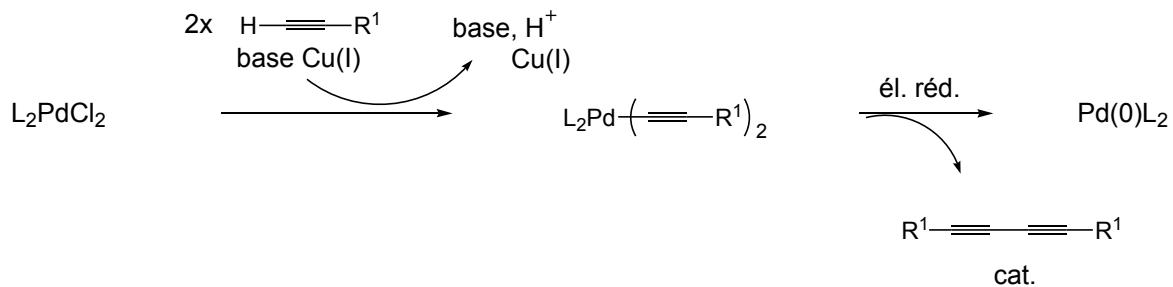
2000ACR2632



## Mécanisme



si utilisation de Pd(II)

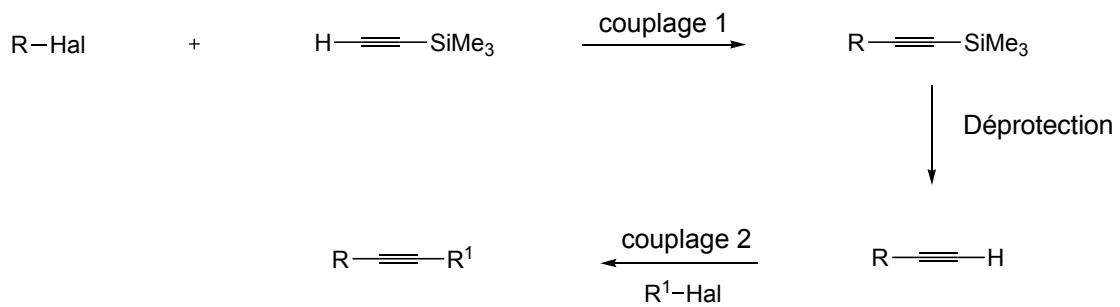


→ consommation de 2 équivalents catalytiques de l'alcyne terminal

Conditions typiques :

PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, CuI, NEt<sub>3</sub>, THF, 25°C-reflux

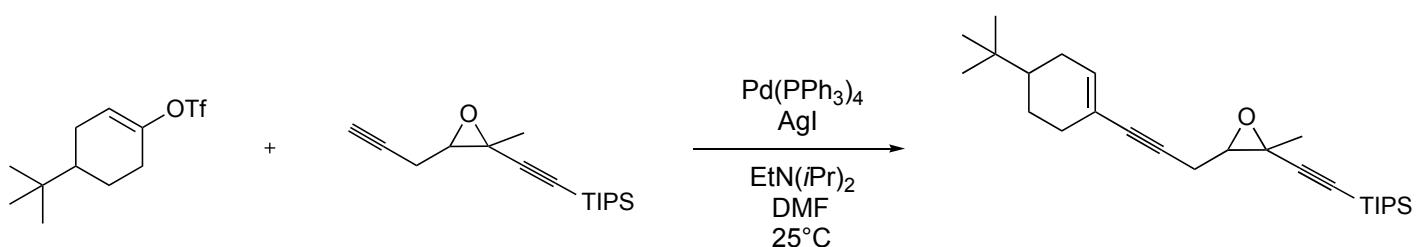
- Bases : souvent amines III
- Synthèse d'alcynes disubstitués



Déprotection in situ avec K<sub>2</sub>CO<sub>3</sub>

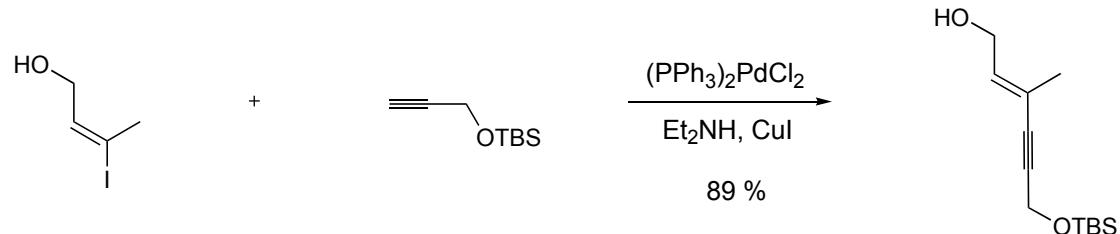
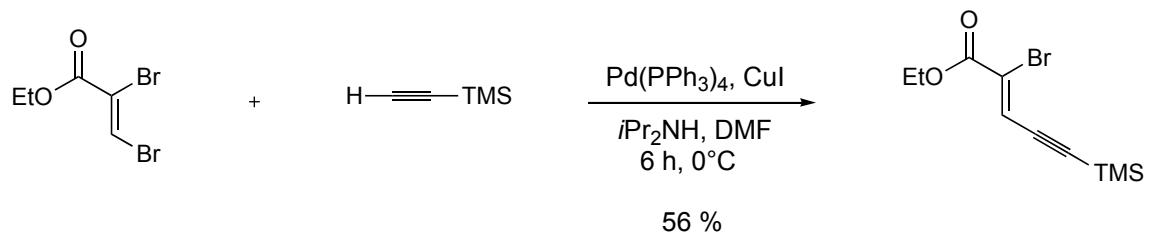
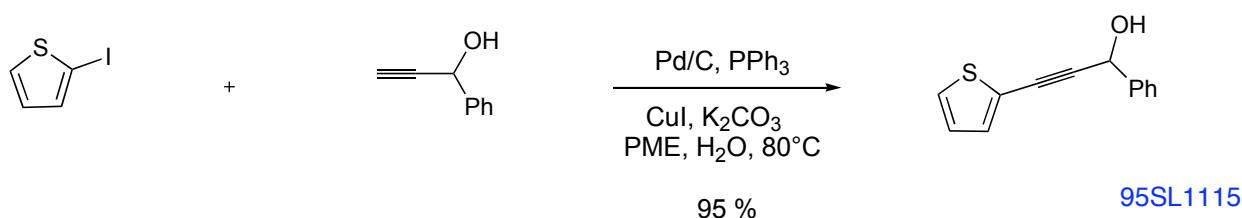
97TL7483

## Exemples

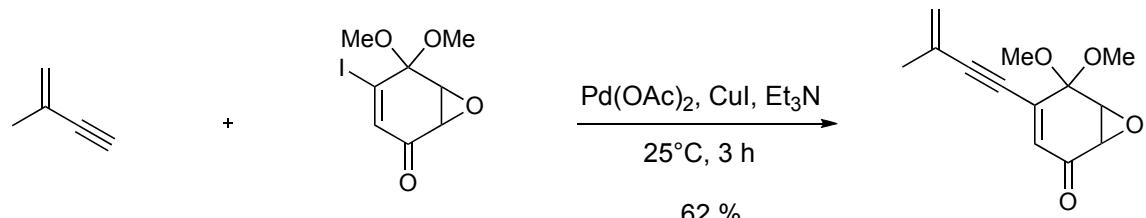


Conditions très douces !  
90 %

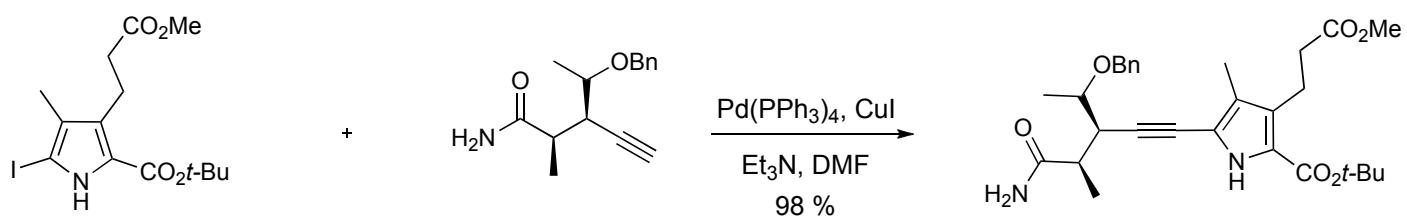
96TL2019



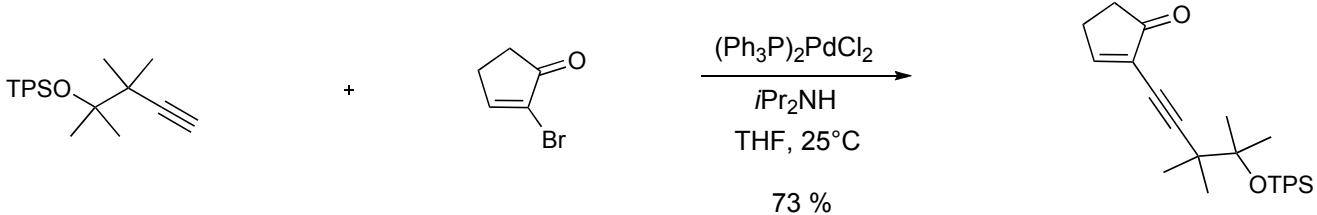
Marshall 93JOC7180



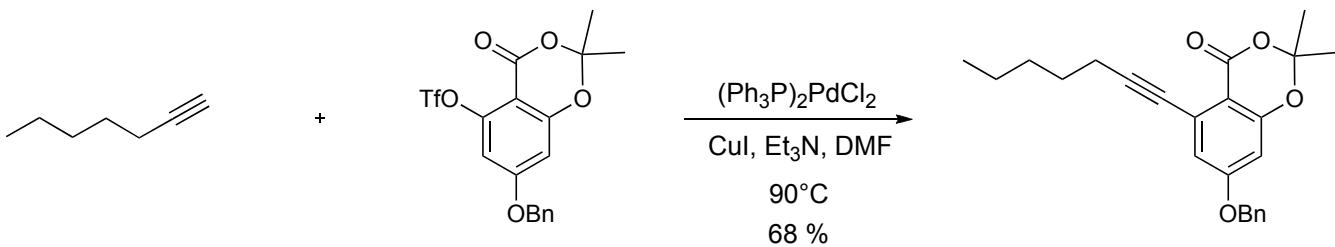
96TL7445



95TL1197

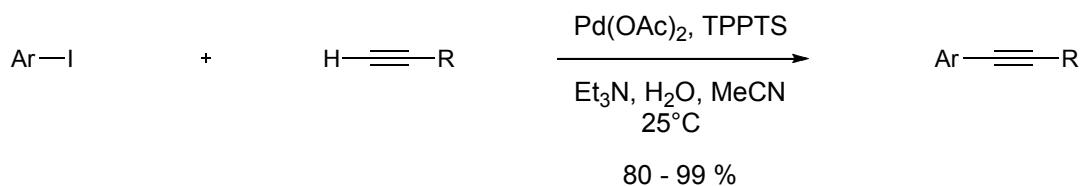


94SC2461



Danishefsky 92JACS655

## Catalyse dans l'eau

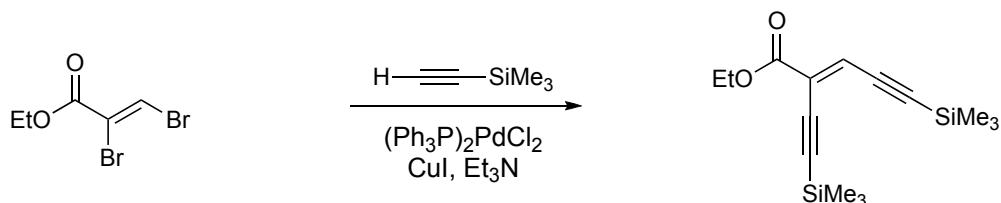
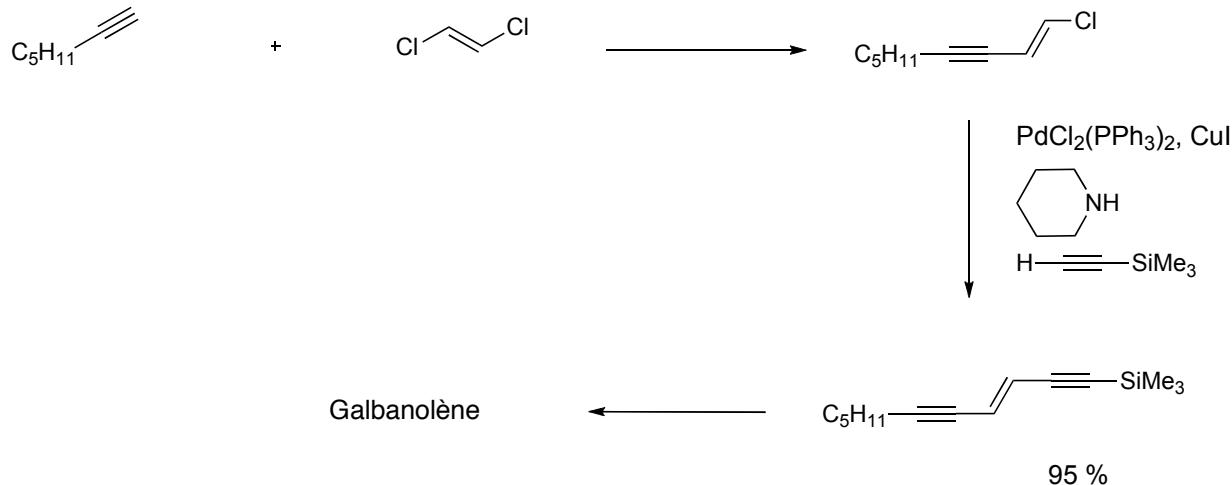


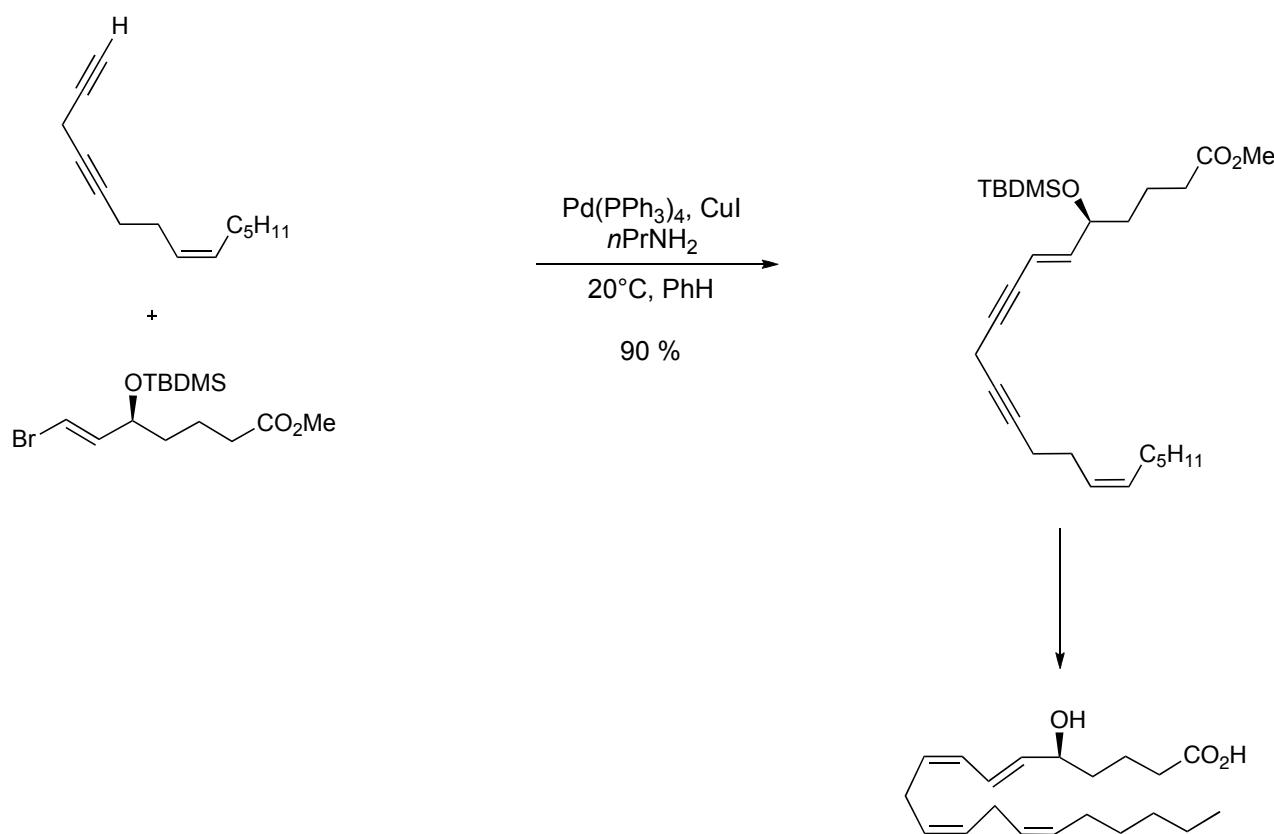
- Aspect de "chimie verte"
- Autre intérêt : pas d'utilisation de Cu

Genêt 92SL715

TPPTS :  $\text{P}(m\text{-C}_6\text{H}_4\text{SO}_3\text{Na})_3$ 

## Application à la synthèse de produits naturels





Nicolaou 86S344

